Flemiphyllin triacetate (2). Flemiphyllin (7 mg) was dissolved in 0.7 ml of Ac_2O , pyridine (3 drops) added and the mixture gently refluxed on an oil bath for 6 hr and poured into crushed ice (50 g) while shaking. After 2 hr the product was filtered, washed throughly with H_2O and crystallized from petrol- C_6H_6 , mp 117–118° (Found: C, 71.85; H, 6.68, $C_{36}H_{40}O_8$ requires C, 71.99; H, 6.66%).

Acknowledgements—We thank Professor K. V. N. Rao, Head of the Department of Botany, Osmania University, for his encouragement and keen interest in the work. We are grateful to U.G.C., New Delhi, India, for financial assistance under grant No. 042/Bio. Sos. 75. One of us (K. N. Rao) is also grateful to C.S.I.R., New Delhi, for awarding a Senior Research Fellowship. We are grateful to Prof. C. V. Ratnam, Head of the Department of Chemistry, for providing the necessary research facilities.

REFERENCES

- Nageswara Rao, K. and Srimannarayana, G. (1983) Phytochemistry 22, 2287.
- 2. Shinoda, J. (1928) J. Pharm. Soc. (Japan) 48, 214.
- Mabry, T. J., Markham, K. R. and Thomas, M. B. (eds) (1970)
 The Systematic Identification of Flavonoids, p. 41 and 253.
 Springer, Berlin.
- 4. Massicot, J., Marathe, J. P. and Heitz, S. (1963) Bull. Soc.

- chim. Fr. 2712.
- 5. Jackman, L. M. (1965) Prog. Chem. Org. Nat. Prod. 23, 315.
- Markham, K. R. and Mabry, T. J. (1975) in *The Flavonoids* (Harborne, J. B., Mabry, T. J. and Mabry, H., eds) pp. 45-75. Chapman & Hall, London.
- Komatsu, M., Tomimori, T., Hatayama, K., Makiguchi, Y. and Mikuriya, N. (1969) Chem. Pharm. Bull (Tokyo) 17, 1299.
- 8. Hillis, W. E. and Horn, D. H. S. (1965) Aust. J. Chem. 18, 531.
- 9. Horowitz, R. M. and Gentili, B. (1964) Chem. Ind. 498.
- Markham, K. R., Mabry, T. J. and Swift, T. W. III (1968) Phytochemistry 7, 803.
- 11. Gibbs, H. D. (1927) J. Biol. Chem. 72, 649.
- Ritchie, E., Taylor, W. C. and Shannon, J. S. (1964) Tetrahedron Letters 1437.
- Deshpande, V. H., Rama Rao, A. V., Malavardan and Venkataraman, K. (1975) Indian J. Chem. 11, 518.
- Mabry, T. J. and Markham, K. R. (1975) in *The Flavonoids* (Harborne, J. B., Mabry, T. J. and Mabry, H., eds) pp. 78-126. Chapman & Hall, London.
- Pelter, A., Bradshaw, J. and Warren, R. F. (1971) Phytochemistry 10, 835.
- Ollis, W. D. and Sutherland, I. O. (1961) in Recent Developments in the Chemistry of Natural Phenolic Compounds, p. 74. Pergamon Press, New York.
- Komatsu, M., Tomimori, T., Hatayama, K. and Mikuriya, N. (1970) Yakagaku Zasshi 90, 463.

Phytochemistry, Vol. 23, No. 4, pp. 929-931, 1984. Printed in Great Britain.

0031-9422/84 \$3.00 + 0.00 © 1984 Pergamon Press Ltd.

BROUSSONETINE, A BISQUINOLYL-γ-BUTYROLACTONE FROM BROUSSONETIA ZEYLANICA*

A. A. LESLIE GUNATILAKA,† SIVAGNANASUNDRAM SURENDRAKUMAR†‡ and RONALD H. THOMSON‡

†Department of Chemistry, University of Peradeniya, Peradeniya, Sri Lanka; ‡Department of Chemistry, University of Aberdeen, Old Aberdeen AB9 2UE, U.K.

(Received 23 September 1983)

Kew Word Index-Broussonetia zeylanica; Moraceae; wood; quinoline alkaloid; broussonetine.

Abstract—The wood of B. zeylanica (Moraceae) contains a new alkaloid broussonetine, identified as 3,4-bis(8-hydroxyquinolin-4-yl)-\(\gamma\)-butyrolactone.

The only Broussonetia species occurring in Sri Lanka is B. zeylanica (Thw.) Corner (= Allaeanthus zeylanicus Thw.) which is endemic to the country. The wood contains 4-formyl-8-hydroxyquinoline (1) [1] and 3,4'-dihydroxy-2,3'-bipyridine (2) [2]. Further investigation has revealed another minor quinoline alkaloid, broussonetine (3).

*Studies on Medicinal and Related Plants of Sri Lanka, Part 12. For part 11, see Gunatilaka, A. A. L., de Silva, A. M. Y. J., Sotheeswaran, S., Balasubramaniam, S., and Wazeer, M. I. M. (1984) Phytochemistry 23, 323.

Broussonetine, $C_{22}H_{16}N_2O_4$, is soluble in dil. hydrochloric acid and dil. sodium hydroxide (yellow), gives positive tests with ferric chloride and Dragendorff's reagent and, like 8-hydroxyquinoline, forms a fluorescent complex with Mg^{2+} ions [3]. The UV spectrum shows λ_{max} at 252 and 333 nm. The presence of two 8-hydroxyquinoline moieties in the alkaloid is evident from the 13 C NMR spectrum; this reveals signals for 18 aromatic carbons which can be assigned as shown (Table 1) and bear a close resemblance to those of 1. The 1 H NMR spectrum (DMSO- d_6) includes a 2H singlet at δ 9.71 (2 × OH), and in the aromatic region two overlapping doublets at δ 8.89 (H-2' and H-2", J = 4.6 Hz) coupled to

930 Short Reports

Table 1.	13C	chemical	shifts*	and	multiplicities	for	compounds	1,	3	and
			8-hv	drox	vauinoline (5)					

1		3		5		
C-2	148.08 d	C-2' and C-2"	147.81 d	C-2 148.13 d		
C-3	125.90 d	C-3' and C-3"	112.76 d	C-3 121.80 d		
C-4	153.46 s	C-4' and C-4"	143.13 s, 145.04 s	C-4 136.03 d		
C-4a	124.68 s	C-4'a and C-4"a	125.86 s, 127. 26 s	C-4a 128.83 s		
C-5	114.36 d	C-5' and C-5"	118.89 d, 119.06 d	C-5 117.77 d		
C-6	130.50 d	C-6' and C-6"	127.64 d, 127.85 d	C-6 127.52 d		
C-7	111.70 d	C-7' and C-7"	111.06 d, 111.21 d	C-7 111.30 d		
C-8	157.16 s	C-8' and C-8"	153.60 s, 153.66 s	C-8 153.32 s		
C-8a	137.09 s	C-8'a and C-8"a	138.57 s, 136.60 s	C-8a 138.49 s		
C-9	193.02 d					
		C-1	174.90 s			
		C-2	36.79 t			
		C-3	42.43 d			
		C-4	80.45 d			

*At 90.56 MHz in DMSO-d₆.

doublets at 7.97 (H-3", $J=4.6\,\mathrm{Hz}$) and 7.70 (H-3', $J=4.6\,\mathrm{Hz}$), respectively, double doublets at 7.09 (H-5' or 5", J=1.0 and 8.60 Hz) and 7.01 (2H, H-7' and H-7", J=1.0 and 7.40 Hz), and complex signals between 7.22 and 7.31 (H-5' or 5", H-6' and H-6"). Extensive homonuclear decoupling of the 360 MHz ¹H spectrum established the presence of two sets of two and three adjacent protons showing that the 8-hydroxyquinoline moieties are substituted at either C-2, C-4, C-5 or C-7. That the 'substituents' are located at C-4' and C-4" in 3 follows from the H-2', H-2" and H-3', H-3" assignments, above, supported by the presence of the downfield doublet at δ 147.81 (2C) in the off-resonance proton decoupled ¹³C NMR spectrum which clearly arises from C-2' and C-2", and from the observation that in the ¹H NMR spectrum of brousso-

netine diacetate (4) the signals from the protons *ortho*- and *para*- to the acetoxyls are all shifted downfield.

The remainder of the molecule is a fragment $C_4H_6O_2$ which must be a γ -butyrolactone unit (ν_{CO} 1775 cm⁻¹). In the ¹H NMR spectrum it shows signals at δ 6.71 (H-4, d, J = 7.4 Hz), 4.84 (H-3, dt, J = 7.4 and 8.6 Hz), 3.33 (H-2a, dd, J = 8.6 and 17.3 Hz) and 3.00 (H-2b, dd, J = 8.6 and 17.3 Hz) consistent with the structure 3,4-bis(8-hydroxyquinolin-4-yl)- γ -butyrolactone (3). In agreement the mass spectrum shows a major peak at m/z 171 (89%) corresponding to fragmentation at a in 3. The $J_{3,4}$ value (7.4 Hz) and the downfield shift of H-4 indicate that the quinoline rings are cis to each other (cf. 3,4-diphenyl- γ -butyrolactone [4]).

Biosynthetically broussonetine (3) could arise from two

Ar CHO + Ac CoA
$$\longrightarrow$$
 Ar \longrightarrow COCoA \longrightarrow Ar \longrightarrow COCoA \longrightarrow Ar \longrightarrow COCoA \longrightarrow Ar \longrightarrow OH \longrightarrow Ar CHO \longrightarrow Ar \longrightarrow OH \longrightarrow Ar \longrightarrow OH \longrightarrow Ar \longrightarrow OH \longrightarrow OH

Scheme 1.

molecules of 1 and one of acetyl-CoA as summarized in Scheme 1.

EXPERIMENTAL

Dried, powdered, wood (3.75 kg) of B. zeylanica collected at Hasalaka, Sri Lanka, was successively extracted with hot light petroleum and hot C₆H₆. The C₆H₆ extract (12.0 g) was transferred to a column of silica gel (Merck, 30-70 mesh) and eluted with C₆H₆ containing increasing amounts of CHCl₃. The C₆H₆-CHCl₃ (49:1) eluates yielded (1), mp 155-156° (420 mg), C₆H₆-CHCl₃ (4:1) gave (2), mp 223-224° (130 mg), both identified by comparison with authentic samples previously isolated [2, 3], and C₆H₆-CHCl₃ (3:2) afforded broussonetine (3) as tiny crystals, mp 238-239° (from CHCl₃) (420 mg), $[\alpha]_D = 0^\circ$ (c 0.11; DMF); (Found: C, 71.3; H, 4.5; N, 7.5%; [M]+, 372.111. $C_{22}H_{16}N_2O_4$ requires C, 70.5; H, 4.3; N, 7.5%; $[M]^+$, 372.1110); UV λ_{max}^{MeOH} nm (log ϵ): 252 (4.19) and 333 (3.70); $\lambda_{max}^{MeOH-NaOH}$ nm: 271 (4.29), 344 (3.99) and 386 (4.07); IR v_{max} r_{max} 13300, 1775, 1570, 1510, 1470, 1405, 1365, 1330, 1270, 1225, 1160, 1130, 1050, 1015, 930, 750 and 690; MS (70 eV) m/z (rel. int.): 372 [M]⁺ (100%), 344 (2), 328 (2), 172 (47), 171.0687 (89, C₁₁H₉NO requires 171.0684), 170 (30), 145 (18), 143 (28), 142 (16), 117 (15). When treated with Ac₂O (1.0 ml) and pyridine (2.0 ml) broussonetine (20 mg) gave a diacetate (4); crystals, mp 162-163° $(C_6H_6-petrol)$ (15.2 mg, 76%). (Found: $[M-CH_2CO]^+$, 414.1245. $C_{24}H_{18}N_2O_5$ requires 414.1215); IR v_{max}^{KBr} cm⁻¹: 1775, 1755, 1601, 1591, 1500, 1470, 1410, 1370, 1310, 1210 br, 1180 br,

941, 771 and 755; ¹H NMR (360 MHz, CDCl₃); δ 2.50 and 2.52 (each 3H, s, OAc), 2.68 (1H, dd, J = 3.1 and 17.9 Hz, H-2a), 3.18 (1H, dd, J = 9.3 and 17.9 Hz, H-2b), 4.50 (1H, dt, J = 3.1 and 9.3 Hz, H-3), 6.51 (1H, dd, J = 0.7 and 3.1 Hz, H-4), 7.12–7.64 (8H, m, ArH), 8.97 and 9.00 (each 1H, d, J = 4.5 Hz, H-2 and H-2'); MS (20 eV) m/z (rel. int.): 456 [M] + (5%), 414 (18), 372 (100), 172 (37), 171 (67), 170 (26), 145 (14), 143 (21) and 117 (7).

Acknowledgements—We thank the International Foundation for Science (Sweden) for financial assistance, Edinburgh University WH-360 NMR Service for NMR spectra, and Professor S. Balasubramaniam for identification of plant material. One of us (S.S.) is grateful to the Leverhulme Trust for an Overseas Visiting Fellowship.

REFERENCES

- Gunatilaka, A. A. L., Perera, J. S. H. Q., Sultanbawa, M. U. S., Brown, P. M. and Thomson, R. H. (1979) J. Chem. Res. (S) 61; (M) 779.
- Gunatilaka, A. A. L., Sultanbawa, M. U. S., Surendrakumar, S. and Somanthan, R. (1983) Phytochemistry 22, 2847.
- Feigl, F. (1960) Spot Tests in Organic Analysis, 6th edn, p. 205. Elsevier. Amsterdam.
- Mandel'shtam, T. V., Kolesova, S. V., Polina, T. V., Solomentsev, V. V. and Osmolovskaya, N. S., (1980) J. Org. Chem. USSR 16, 1024.

Phytochemistry, Vol. 23, No. 4, pp. 931-933, 1984. Printed in Great Britain.

0031-9422/84 \$3.00+0.00 © 1984 Pergamon Press Ltd.

SATIVANINE-C: A CYCLOPEPTIDE ALKALOID FROM THE BARK OF ZIZYPHUS SATIVA*

A. H. Shah, V. B. Pandeyt, G. Eckhardt and R. Tscheschet

Department of Chemistry, Gomal University, D. I. Khan, Pakistan; †Department of Medicinal Chemistry, IMS, B.H.U., Varanasi, India; ‡Institut für Organische Chemie und Biochemie der Universität, Gerhard-Domagk Str 1. D-5300 Bonn 1. W. Germany

(Received 18 July 1983)

Key Word Index-Zizyphus sativa; Rhamnaceae; cyclopeptide alkaloid; sativanine-C.

Abstract—From the bark of Zizyphus sativa a previously undescribed 13 membered cyclopeptide alkaloid, sativanine-C has been isolated. The structure of this new compound was elucidated by spectroscopic methods, its transformation product and by chemical degradation.

INTRODUCTION

As a part of our extended studies on the alkaloids of the Rhamnaceae we recently reported the isolation and characterization of five cyclopeptide alkaloids from Zizyphus sativa [2]. Extensive chromatography of the

crude bases furnished a further previously unknown 13-membered cyclopeptide alkaloid (1). Its structure is related to nummularine-B (3) [3].

RESULTS AND DISCUSSION

The alkaloid was isolated from the polar fraction by TLC on silica gel. The molecular formula was determined by high resolution mass spectrometry as C₂₉H₄₃N₅O₆. The IR spectrum displayed characteristic secondary am-

^{*}Part 34 in the series "The Alkaloids of Rhamnaceae". For Part 33 see ref. [1].